

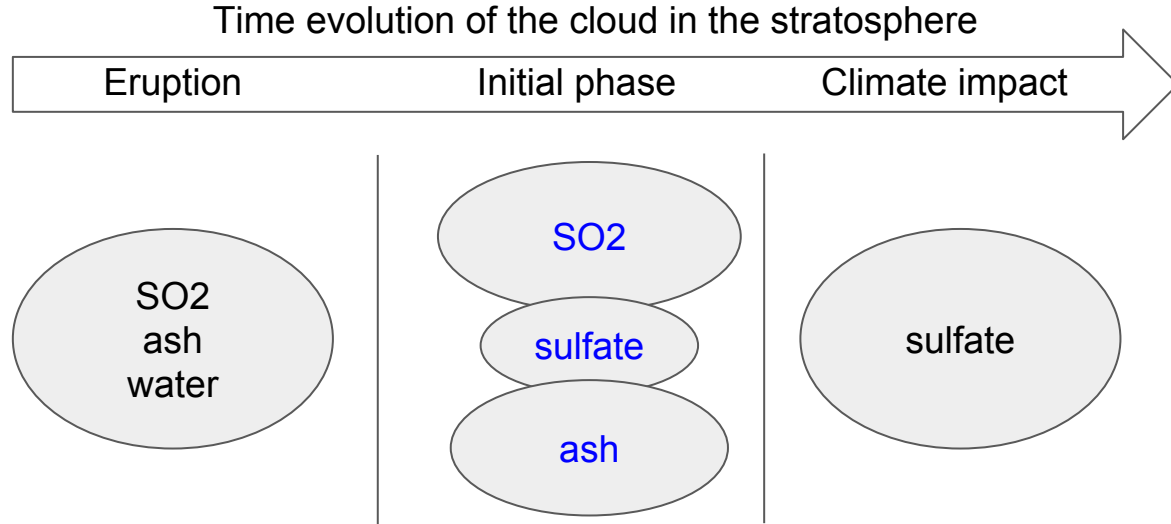
# The effects of SO<sub>2</sub>, volcanic ash and sulfate aerosols on photolysis rates and the sulfate chemical production following the volcanic eruptions

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SO<sub>2</sub> key points:

1. greenhouse gas effect
2. SO<sub>2</sub>-sulfate cloud heating
3. photolysis rates reduction

# Introduction



The evolution of the  $\text{SO}_2$ -sulfate cloud depends on the atmospheric **dynamics**, **photochemistry**, and microphysics.

We account that  $\text{SO}_2$ , sulfate and ash are optically active in SW and LW including the UV range.

**Pinatubo, 1991, 20Mt of  $\text{SO}_2$**

8 ppmv, 400 km x 2000 km x 14-21 km

**Toba, 75 000 years ago, 100x Pinatubo, 2000 Mt of  $\text{SO}_2$**

15.5 ppmv, equatorial belt 15S-15N x 10-50 hPa

# Modelling setup

## Standalone models:

1. Line-by-line AER LBLRTM coupled to DISORT
2. Column AER SW/LW RRTM/RRTMG with added SO<sub>2</sub> effects (thanks to Mike Iacono, Eli Mlawer, Karen Cady-Pereira)

## Initial phase

NCAR WRF Chem v3.7.1:

Global 1°x1° setup, spectral U&V nudging

Prescribed SST

Interactive chemistry: GOCART(sulphur cycle) coupled with RACM-KPP, fixed climatological ozone

Aerosols are coupled to the chemistry.

Modifications:

1. SO<sub>2</sub> included in RRTMG SW&LW (AER custom version, *dynamics*)
2. SO<sub>2</sub> included in Fast-j actinic flux (*photochemistry*)
3. Double radiation and chemistry calls for instantaneous online diagnostics
4. Volcanic emissions added though the IC or inventory

## Climate impact

NASA GISS ModelE2:

Global 2°x2.5° setup

Coupled ocean (Russel) and sea-ice models

Interactive chemistry (Shindell et al. [2006])

Aerosols are coupled to the chemistry.

Modifications (future work):

1. SO<sub>2</sub> effect in SW (currently only in LW)
2. SO<sub>2</sub> effect in Fast-j actinic flux

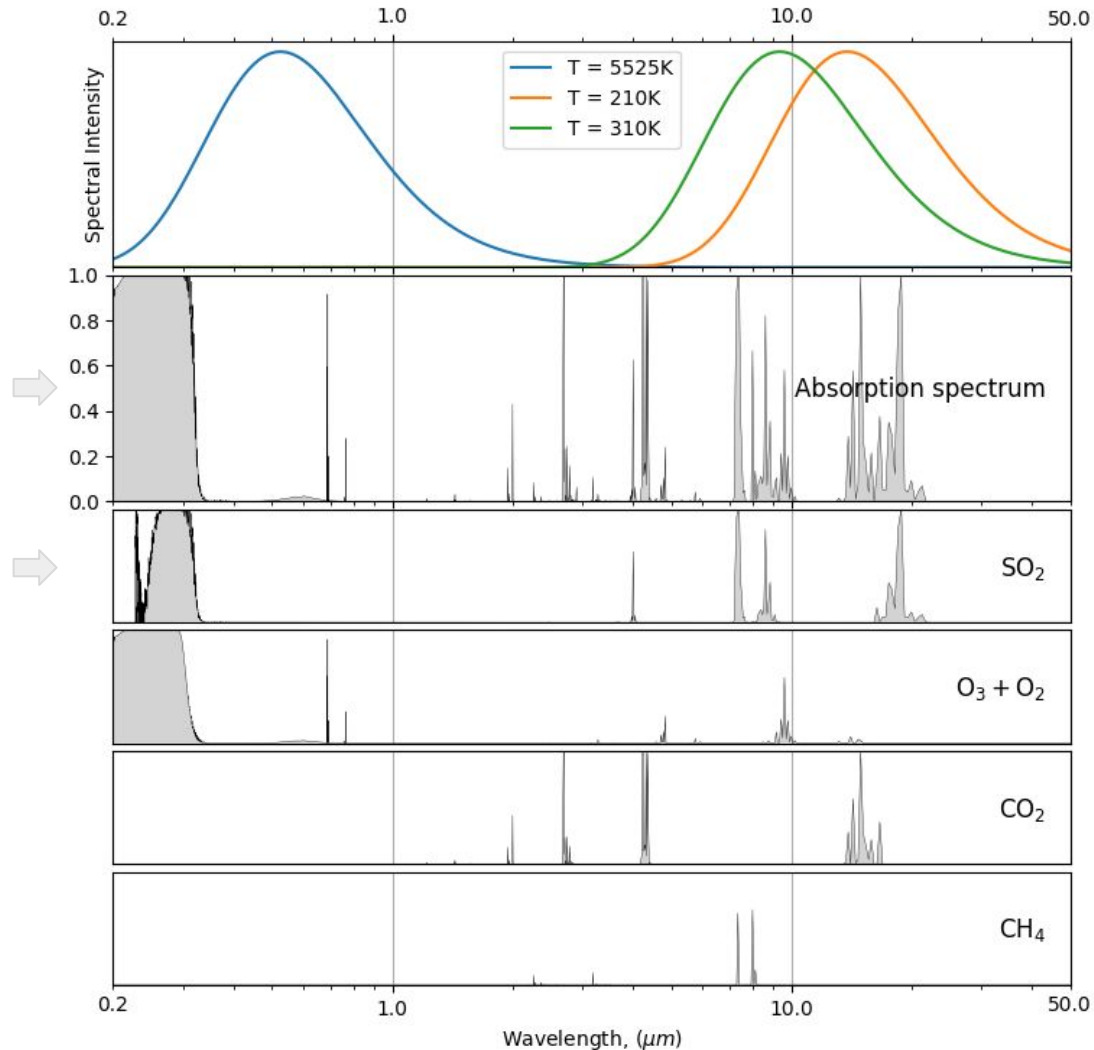
# Motivation

10-50 hPa line-by-line absorption spectrum.  
Realistic equatorial atmosphere + Toba eruption (15.5 ppmv  $\text{SO}_2$ )

$\text{SO}_2$  is a strong greenhouse gas.  
 $\text{SO}_2$  competes with  $\text{O}_3$  in UV absorption.

## $\text{SO}_2$ effects:

1. Radiative forcing
2. Dynamics
3. Photochemistry



# SO<sub>2</sub> radiative forcing, Toba, Wm<sup>-2</sup>

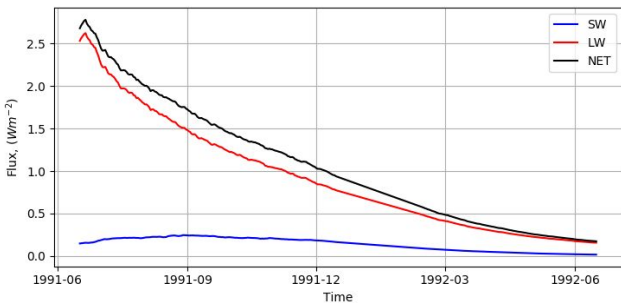
Single profile (RRTM column model)  
15.5 ppmv of SO<sub>2</sub>, 10-50 hPa

TOA	SW	0.69	BOA	SW	-1.08
	LW	12.66		LW	0.32
	NET	13.35		NET	-0.77

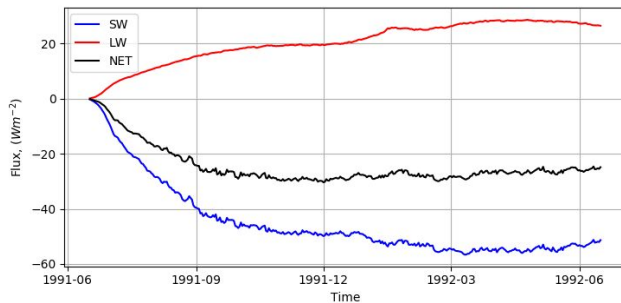
Global average (WRF-Chem)  
15.5 ppmv of SO<sub>2</sub>, 10-50 hPa, 15S-15N

TOA	SW	0.16	BOA	SW	-0.12
	LW	2.62		LW	0.03
	NET	<b>2.78</b>		NET	-0.08

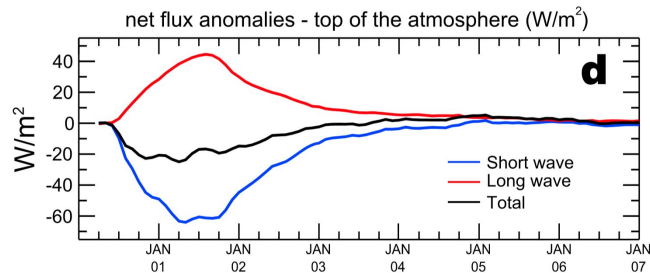
SO<sub>2</sub> TOA forcing  
WRF-Chem



sulfate TOA forcing  
small  $r_{\text{eff}}$ , WRF-Chem



sulfate TOA forcing  
large  $r_{\text{eff}}$ , Timmreck et. al., 2010



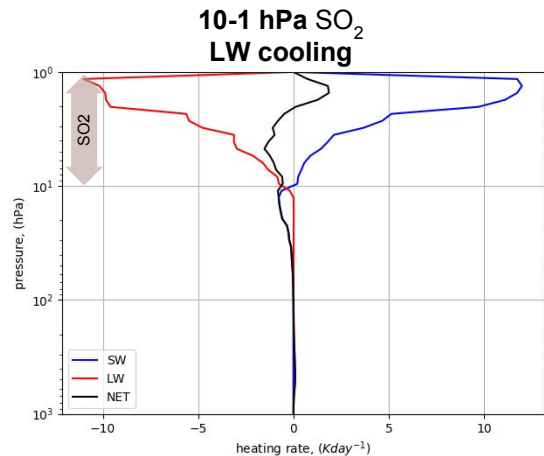
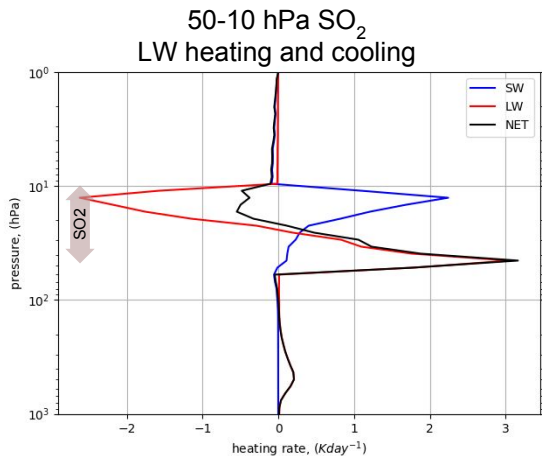
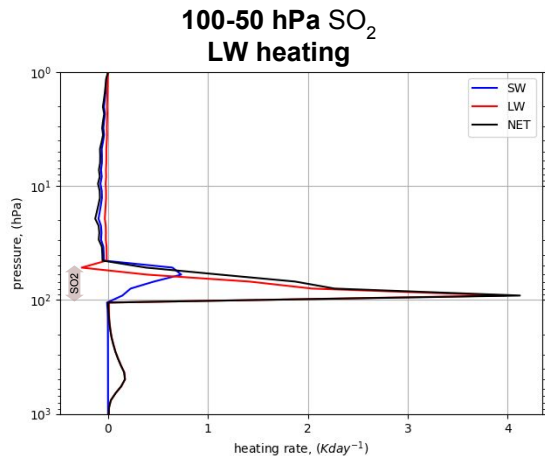
Max **NET TOA forcing** of SO<sub>2</sub> represents **10-15%** of sulfate forcing (2.8 Wm<sup>-2</sup> vs 20-30 Wm<sup>-2</sup>)

# SO<sub>2</sub> effect on dynamics: heating rates

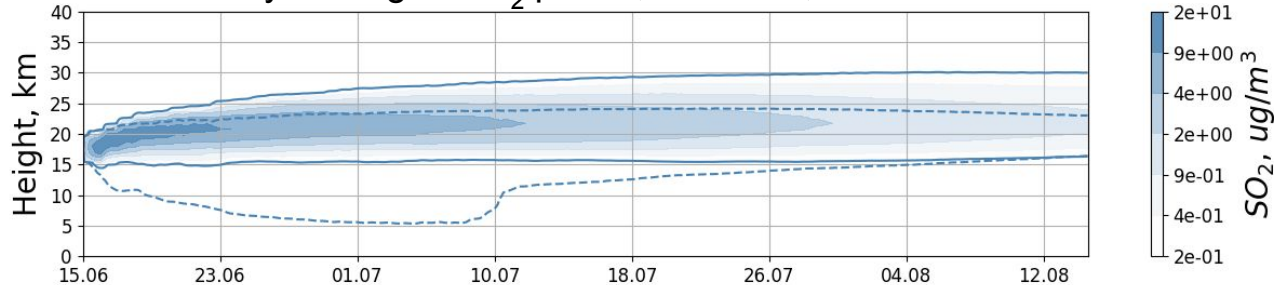
Poster P-28

SO<sub>2</sub> daily mean heating rates, column RRTM model, Toba

Zhong et. al., 1996



Globally averages SO<sub>2</sub> profile, Pinatubo, WRF-Chem

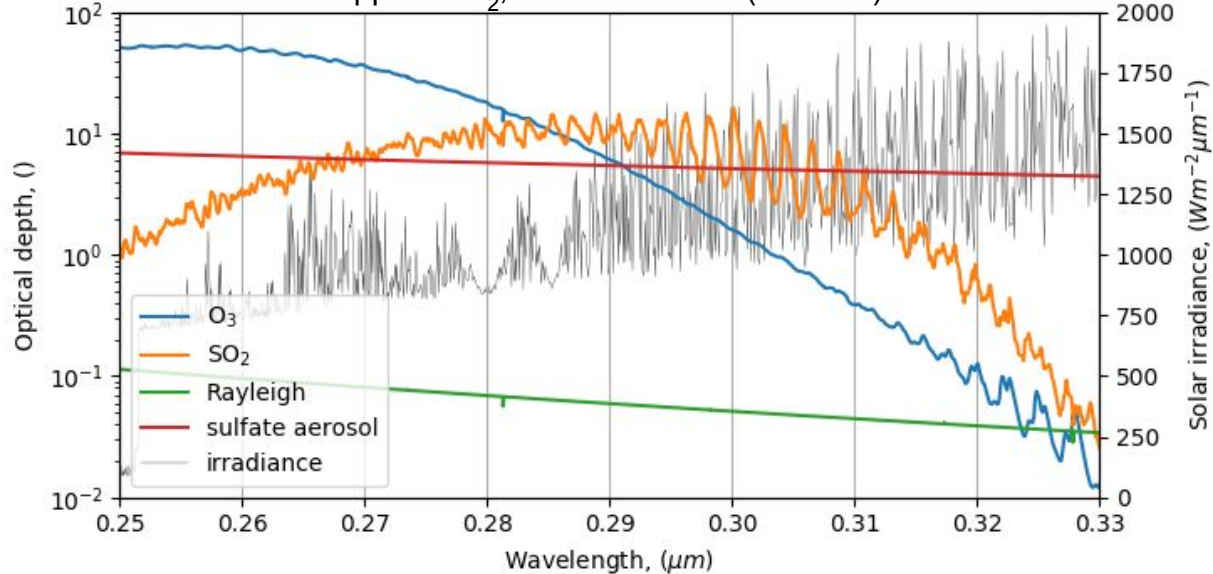


Radiative feedback on dynamics  
solid lines: on  
dashed lines: off

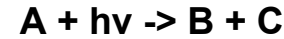
**Different cloud evolution and transport due to SO<sub>2</sub> heating rates**

# SO<sub>2</sub> effect on the photochemistry: UV spectrum and O<sub>3</sub>

UV spectral optical depth, 10-50 hPa, LBLRTM  
15.5 ppmv SO<sub>2</sub>, sulfate column  $\tau(0.55 \mu\text{m})=2.5$



← O<sub>3</sub> photolysis relevant range,  $\phi > 0$        $\phi = 0$  →



$$\frac{d[A]}{dt} = -J_A[A] = -\int \sigma_A(\lambda)\phi_A(\lambda)F(\lambda)d\lambda \times [A]$$

$\sigma$  - absorption cross section of A

$\phi$  - quantum yield

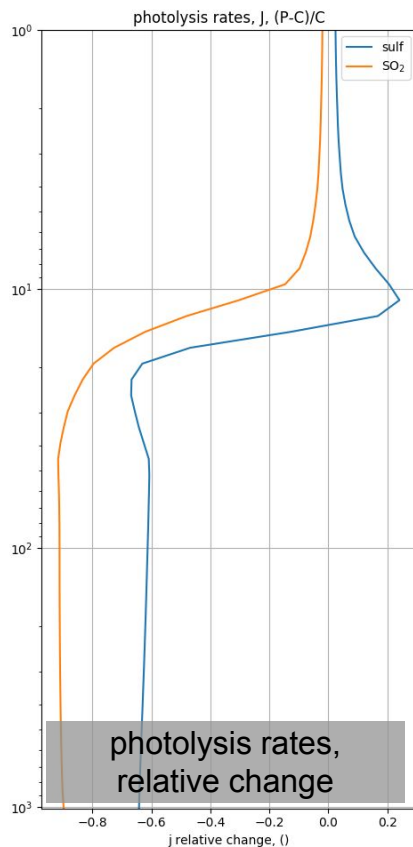
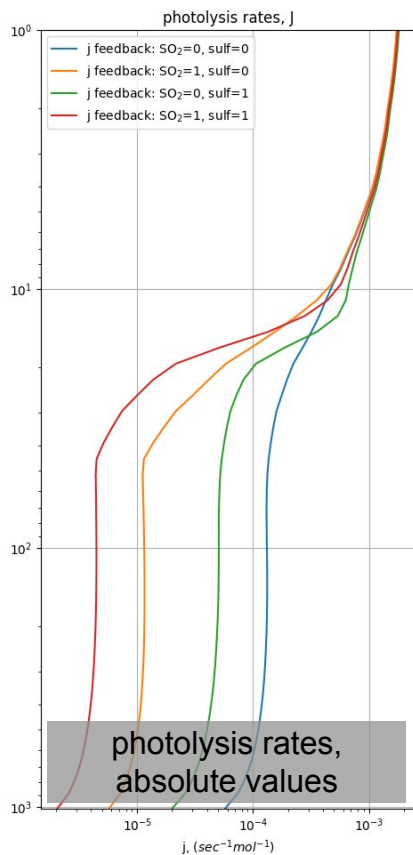
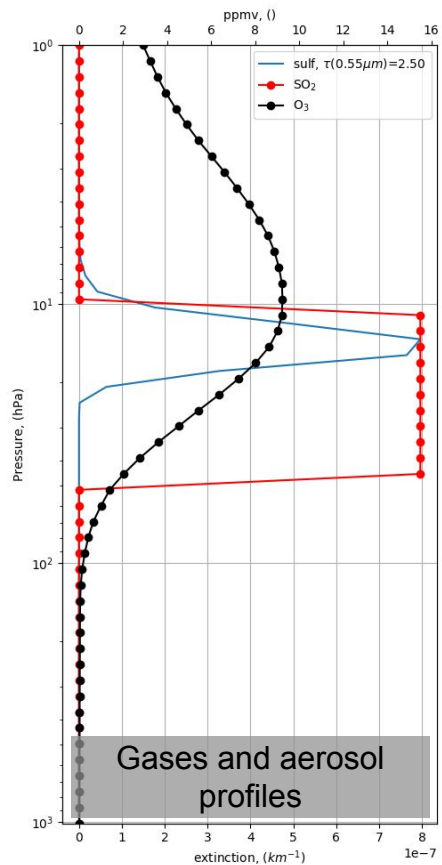
$F$  - actinic flux

O<sub>3</sub> photolysis is a primary source of OH, which drives the SO<sub>2</sub> oxidation.

**Actinic flux reduction due to SO<sub>2</sub> absorption** reduces O<sub>3</sub> photolysis rates, SO<sub>2</sub> depletion and sulfate production

# SO<sub>2</sub> and sulfate effect on photolysis rates J, Toba

daylight mean photolysis rates, LBLRTM+DISORT



OH is a short-lived radical. Changes in photolysis rates J are only relevant for sulfate production where SO<sub>2</sub> is present.

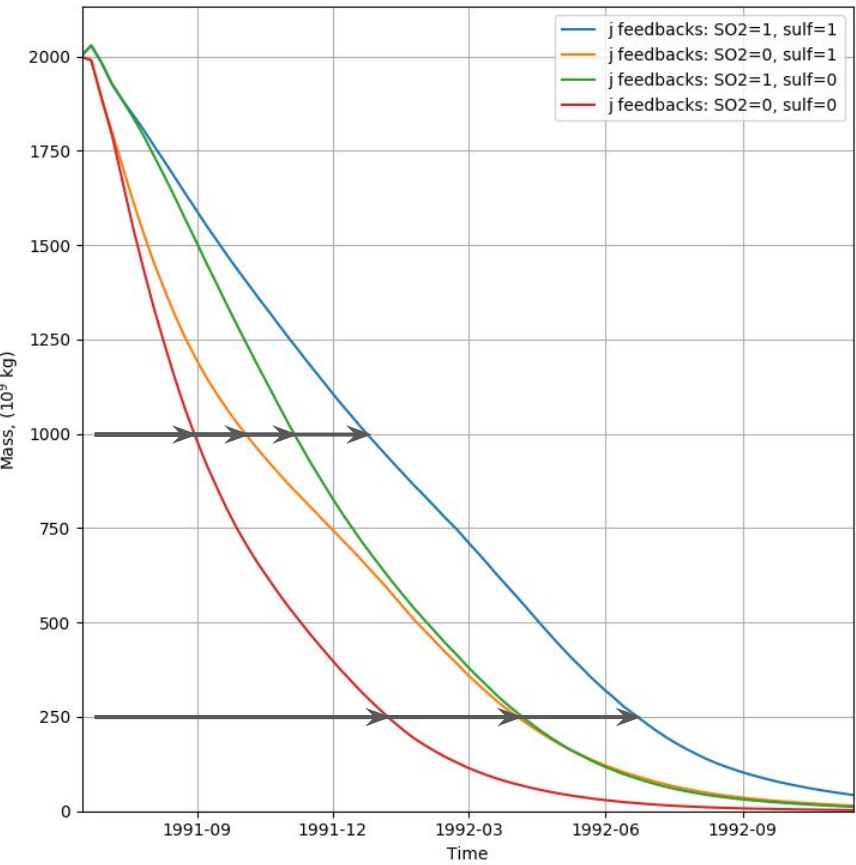
relative change = (P-C)/C  
P: both SO<sub>2</sub> and sulf  
C: only one (SO<sub>2</sub> or sulfate)

J relative change due to:  
SO<sub>2</sub>: -90% to -20%  
sulfate: -70% to 20%



# SO<sub>2</sub> depletion time evolution, Toba

Globally integrated SO2 mass, WRF-Chem

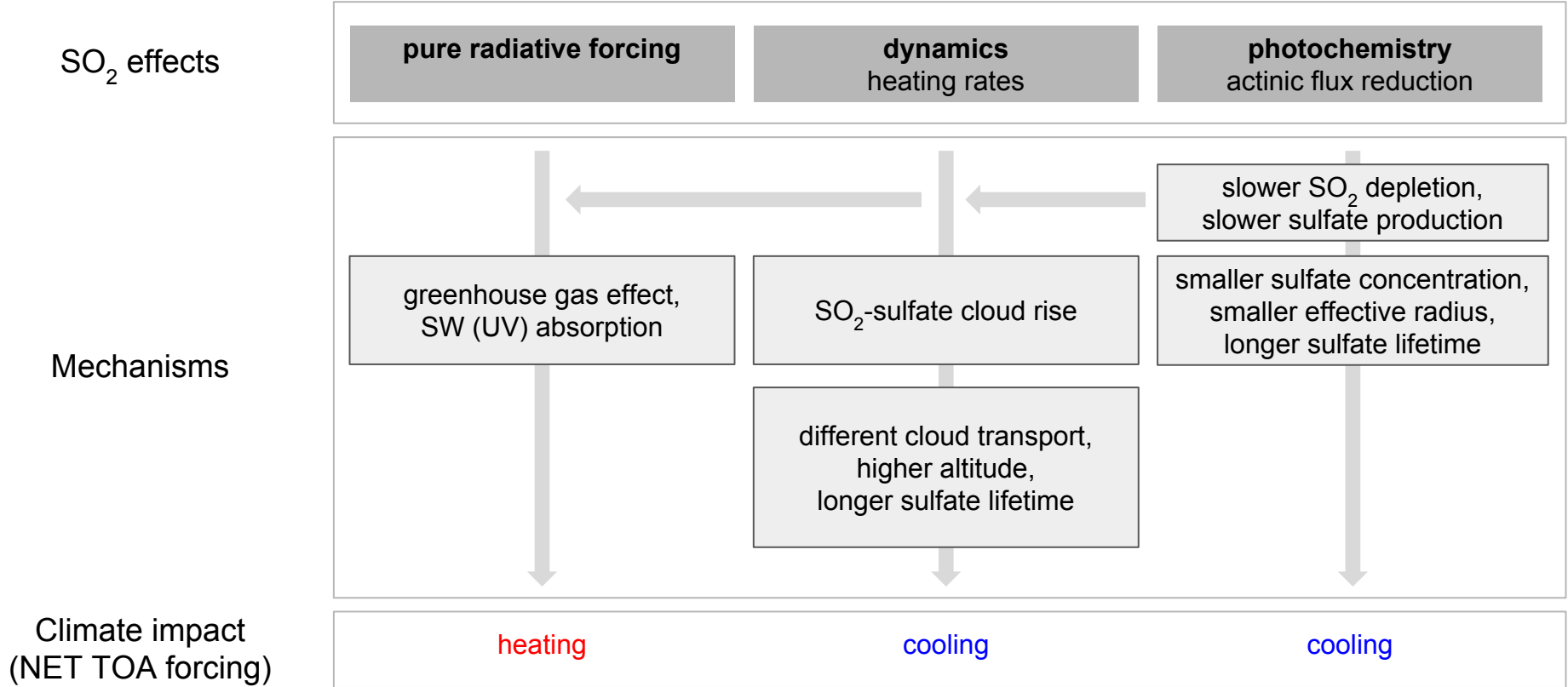


SO<sub>2</sub> time scales  
w.r.t experiment without feedbacks

interval	-	sulf	SO <sub>2</sub>	sulf, SO <sub>2</sub>
e-time (1st month)	x1 4 months	x1.15 4.6 months	<b>x2.3</b> 9.2 months	x2.3 9.3 months
50% SO <sub>2</sub>	x1 2.6 months	x1.3 3.7 months	<b>x1.6</b> 4.9 months	x2.3 6.5 months
12.5% SO <sub>2</sub>	x1 7 months	x1.4 9.8 months	<b>x1.4</b> 10 months	x1.9 12.6 months

**1.4 - 2.3 times slower SO<sub>2</sub> depletion and sulfate production** due to effect on photochemistry

# Conclusions, Toba



# Conclusions, Pinatubo

